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4.3 reaction stoichiometry chemistry 2e openstax Dec 10 2020 solution the approach used previously in example 4.8 and example 4.9 is likewise used here that is we must derive an appropriate stoichiometric factor from the balanced chemical equation and use it to relate the amounts of the two substances of interest in this case however masses not molar amounts are provided and requested so additional steps of the sort learned in the

the physics classroom tutorial Mar 25 2022 compared to the previous problem this is a much more difficult problem in fact this problem is like two problems in one at the center of the problem solving strategy is the recognition that the quantity of heat lost by the water q_{water} equals the quantity of heat gained by the metal q_{metal} since the m_c and Δt values of the water are known the q_{water} can be calculated

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